

Year 10 Combined Science	T1	T2	Т3	T4	T5	Т6
Content / Topic for Term	Year 9 content recap Quantitative chemistry	Structure and bonding	Structure and bonding (cont)	Chemical changes (cont)	Chemical changes (cont)	Revisit Year 9 knowledge and skills
Key Knowledge for acquisition, recall and application in assessment or exam	Recap Year 9 content • First three weeks review and recap key Year 9 content Quantitative chemistry • balancing equations • relative masses and moles • mass moles • mass moles • calculations • from masses to balanced equations	Structure and bonding • sates of matter • chemical bonds • ionic bonding • ionic compounds • covalent bonding • simple covalent structures • giant covalent bonding	Structure and bonding (cont) • metallic bonding • bonding summary • fullerenes and graphene Chemical changes • reactivity series • displacement reactions • reactions of metals	Chemical changes (cont) • extracting metals • acids and bases • naming salts • making salts required practical • strong and weak acids • electrolysis • electroplating • electrolysis of aluminium	Chemical changes (cont) • electrolysis of solutions • electrolysis required practical Year 9 topics revisited • atomic structure and the periodic table	Year 9 topics revisited • chemical changes • using resources • chemistry of the atmosphere
Key skills to apply in assessment or exam	 Use scientific vocabulary, terminology and definitions. 	 Use scientific vocabulary, terminology and definitions. 	 Use scientific vocabulary, terminology and definitions. 	 Use scientific vocabulary, terminology and definitions. 	 Use scientific vocabulary, terminology and definitions. 	 Use scientific vocabulary, terminology and definitions.



	 Carrying out and representing mathematical and statistical analysis. Translating data from one form to another. 	 Apply a knowledge of a range of techniques, instruments, apparatus, and materials to select those appropriate to the experiment. Evaluate risks both in practical science and the wider societal context, including perception of risk in relation to data and consequences. 	 Plan experiments or devise procedures to make observations, produce or characterise a substance, test hypotheses, check data or explore phenomena. 	 Presenting reasoned explanations including relating data to hypotheses. Use SI units (eg kg, g, mg, km, m, mm, kJ, J) and IUPAC chemical nomenclature unless inappropriate. 	 Presenting observations and other data using appropriate methods. Explain every day and technological applications of science; evaluate associated personal, social, economic and environmental implications; and make decisions based on the evaluation of evidence and 	
					evidence and arguments.	
Title of Knowledge Organiser	• Quantitative chemistry	EnergyRates of reaction	• Reactions of acids and baes	 Reactions of acids and baes Electrolysis 	 Electrolysis Year 9 atomic structure and the periodic table 	 Year 9 chemical changes Year 9 using resources Year 9 chemistry of the atmosphere



Year 10 Seperate	T1	T2	Т3	T4	Т5	Т6
Science Content / Topic for Term Key	Year 9 content recap Quantitative chemistry Recap Year 9	Quantitative chemistry (cont) Structure and bonding Quantitative	Structure and bonding Structure and bonding (cont)	Chemical changes Chemical changes	Chemical changes (cont) Chemical changes	Chemical changes (cont) Revisit Year 9 knowledge and skills Revisit Year 9 knowledge and
for acquisition, recall and application in assessment or exam	 first three weeks review and recap key Year 9 content Quantitative chemistry balancing equations relative masses and moles mass moles calculations from masses to balanced equations % yield atom economy 	 titrations titration calculations volumes of gases Structure and bonding sates of matter chemical bonds ionic bonding ionic compounds covalent bonding 	 simple covalent structures giant covalent Bonding metallic bonding bonding summary fullerenes and graphene 	 displacement Reactions reactions of metals extracting metals acids and bases naming salts making salts Required Practical 	 strong and weak acids electrolysis electrolysis of aluminium electrolysis of solutions electrolysis 	 skills atomic structure and the periodic table chemical changes using resources chemistry of the atmosphere



	 concentration calculations 					
Key skills to apply in assessment or exam	 Use scientific vocabulary, terminology and definitions. Carrying out and represent mathematical and statistical analysis. Translating data from one form to another. 	 Use scientific vocabulary, terminology and definitions. Apply a knowledge of a range of techniques, instruments, apparatus, and materials to select those appropriate to the experiment. Evaluate risks both in practical science and the wider societal context, including perception of risk in relation to data and consequences. 	 Use scientific vocabulary, terminology and definitions. Plan experiments or devise procedures to make observations, produce or characterise a substance, test hypotheses, check data or explore phenomena. 	 Use scientific vocabulary, terminology and definitions. Presenting reasoned explanations including relating data to hypotheses. Use SI units (eg kg, g, mg, km, m, mm, kJ, J) and IUPAC chemical nomenclature unless inappropriate. 	 Use scientific vocabulary, terminology and definitions. Presenting observations and other data using appropriate methods. Explain every day and technological applications of science; evaluate associated personal, social, economic and environmental implications; and make decisions based on the evaluation of evidence and arguments. 	Use scientific vocabulary, terminology and definitions.
Title of Knowledge Organiser	 Quantitative chemistry 	 Quantitative chemistry Structure and bonding 	 Structure and bonding 	 Reactions of acids and bases 	 Reactions of acids and bases Electrolysis	• Year 9 atomic structure and the periodic table



			• Year 9 chemical
			changes
			 Year 9 using
			resources
			• Year 9 chemistry of
			the atmosphere